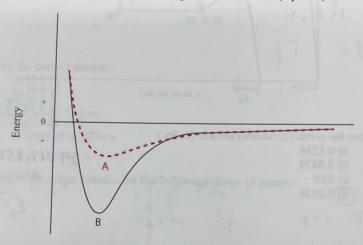
CE 60 PROPERTIES OF CIVIL ENGINEERING MATERIALS EXAMINATION

Question 1 (30 points)

I) An energy versus interatomic distance plot is shown below. (True False) Curve A has a higher elastic modulus compared with curve B (6 points).



II) The atomic number of Carbon is 6. How many protons, neutrons, and electrons are there in a single atom? (6 points). What is the atomic mass of the atom? Use the unit amu. (2 points). Assume the atom has a net zero charge and no isotopes.

6 protons, 6 electrons, 6 neutrons in a single atom.

atomic mass = # protons + # neutrons

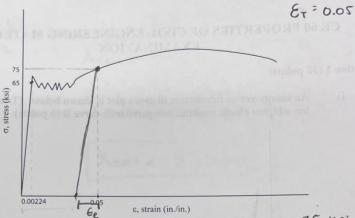
atomic mass = 12 amu

≯ III)

A tension test is performed on a low-carbon steel sample with an initial diameter of 0.5 inches and initial gauge length of 2 inches. The stress-strain curve of the sample is shown below. Calculate the elastic strain at a stress of 75,000 psi. (8 points)

E= = 65 KSi

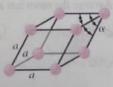
E = 29017.857 HSi



Choose the correct (rounded) answer (in./in.):

- a) 0.1234
- b) 0.0474
- c) 0.04/4
- (d) 0.0026

IV) Identify the following crystal structure: The lattice parameters for this crystal structure are $\mathbf{a}=\mathbf{b}=\mathbf{c}$ and $\alpha=\beta=\gamma\neq 90^{\circ}$. (8 points)

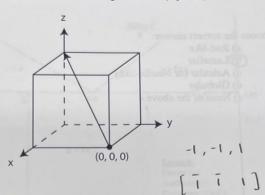


Choose the correct answer:

- a) Simple tetragonal
- b) Triclinic
- (c) Rhombohedral
- d) Simple orthorhombic
- e) None of the above

Question 2 (24 points)

I) Define the Miller indexes for the following direction: (8 points)



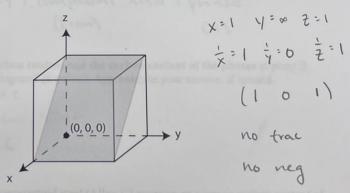
Choose the correct answer:

a) $\begin{bmatrix} 1 & 1 & 1 \\ 0 & (\overline{1} & \overline{1} & 1) \end{bmatrix}$ b) $(\overline{1} & \overline{1} & 1)$ c) $(\overline{1} & \overline{1} & 1]$

d) (1 1 1)

e) None of the above

Define the Miller indexes for the following plane: (8 points)



Choose the correct answer:

a) $(1 \bar{1} 0)$

b) [0 1 0]

©(1 0 1) d) [1 1 1]

e) None of the above

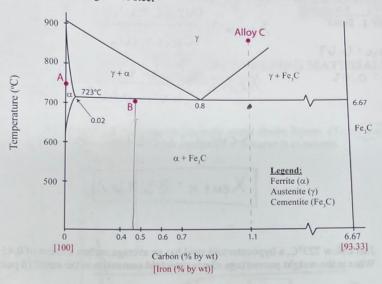
Pearlite is a combination of eutectoid ferrite and eutectoid cementite. What is the morphology of this combination? (8 points) × III)

Choose the correct answer:

- a) Rod-like
 (b) Lamellar
 c) Acicular (or Needle-like)
- d) Globular
- e) None of the above

Question 3: (46 points)

Considering the phase diagram of steel



Using Gibbs phase rule as a guide, how many components and phases are at point A? Assume pressure is constant. (4 points)

Point A nas only I component and I phase

What is the average carbon content and the carbon content of the phases at point B indicated in the phase diagram? (6 points). Approximate your answer, if needed.

Wo = 0.46 wt%. C Wx = D. 8 wt 1. 6 WFe3 C Wx = 0.02 wtil. C

> III) Just below 723°C, a hypereutectoid steel (Alloy C) contains an average carbon content of 1.1 wt %. What is the weight percentage of pearlite? (15 points). Show your work.

Xpeanute = Xeut a + Xeut Fez C Xeut Fe 3 C = X tot Fe 3 C - X profe 5 C Xeut Fe36= 1.1-.02 - 1.1-.8 Wo= 1.1 WFesc= 6.67 Xeut x = . 8376 Wx=0.8 Xent Feac = . 1113

Wa = 0.02

X pearlite = . 9489

hyper

IV) Just below 723°C, a hyppreutectoid steel has an average carbon content of 1.1 %. What is the weight percentage of eutectoid ferrite in the steel? (15 points). Show your work.

 $W_{est} = 0.02$ Xent $\alpha = \frac{W_{est} - W_{o}}{W_{est} - W_{a}}$ $W_{a} = 0.02$ Xent $\alpha = \frac{6.67 - 1.1}{6.67 - 0.02}$

Xeu+ x = .8376

Xeut x = 83.76%

V) Just below 723°C, a hypoeutectoid steel has an average carbon content of 0.45 %. What is the **weight percentage of proeutectoid cementite** in the steel? (6 points)

The steel was hypoentectoid so there is only proentectoid ferrite.